

Complete the following table. **ANSWERS**

Name	Symbol	Atomic Number	Mass Number	Number of Neutrons
oxygen	$^{17}_8\text{O}$	8	17	9
neon	$^{20}_{10}\text{Ne}$	10	20	10
iron	$^{56}_{26}\text{Fe}$	26	56	30
plutonium	$^{244}_{94}\text{Pu}$	94	244	150
mercury	$^{202}_{80}\text{Hg}$	80	202	122
cobalt	$^{59}_{27}\text{Co}$	27	59	32
nickel	$^{56}_{28}\text{Ni}$	28	56	28
fluorine	$^{19}_9\text{F}$	9	19	10
chromium	$^{50}_{24}\text{Cr}$	24	50	26