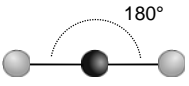
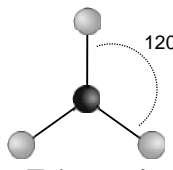
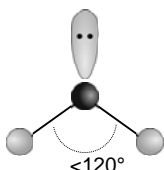
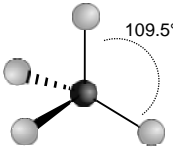
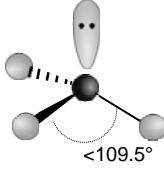
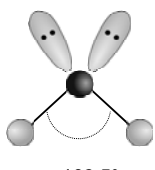
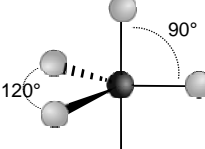
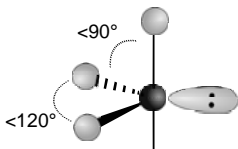
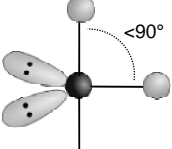
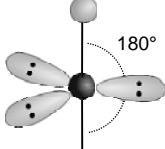
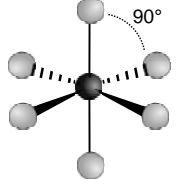
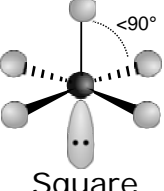
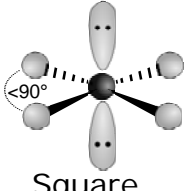
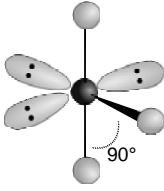
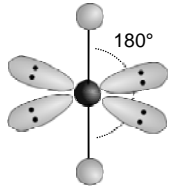


# Valence Shell Electron Pair Repulsion (VSEPR) Geometries

Steric #	(Base Geometry)				
	0 Lone Pairs	1 Lone Pair	2 Lone Pairs	3 Lone Pairs	4 Lone Pairs
2	 <p>Linear</p>				
3	 <p>Trigonal Planar</p>	 <p>Bent</p>			
4	 <p>Tetrahedral</p>	 <p>Trigonal Pyramid</p>	 <p>Bent</p>		
5	 <p>Trigonal Bipyramid</p>	 <p>See-Saw</p>	 <p>T-Shaped</p>	 <p>Linear</p>	
6	 <p>Octahedral</p>	 <p>Square Pyramid</p>	 <p>Square Planar</p>	 <p>T-Shaped</p>	 <p>Linear</p>

**Note:** Any structure with only two bonded atoms only is linear.